

• 研究简报 •

# 新型腙类衍生物的合成及其生物活性

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**摘要:** 利用 3-(N-甲基-N-甲氧基氨基)-1-芳基-1-丙酮腙与酰氯反应, 合成了 14 个新型 N-酰基-3-(N-甲基-N-甲氧基氨基)-1-芳基-1-丙酮腙化合物, 其化学结构经<sup>1</sup>H 核磁共振和元素分析确证。初步的生物活性测试结果表明, 部分化合物在 50 mg/L 时对淡色库蚊 Culex pipiens pallens 的致死率为 95% ~ 100%; 化合物 e 在 1 000 mg/L 时对粘虫 Leucania separata Walk 的致死率达 100%。

**关键词:** 酰化反应; 芳基丙酮腙; 合成; 生物活性

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## Synthesis and Bioactivity of a Series of Novel Hydrazone Derivatives

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**Abstract** In search of novel pesticide compound with high bioactivity, fourteen new N-3-(N-methoxy-N-methylamino)-1-arylpromylidene-hydrazide derivatives have been designed and synthesized by the reaction of N-[3-hydrazono-propyl]O, N-dimethylhydrazine with different acyl chlorides. Their structures have been confirmed by <sup>1</sup>H NMR and elemental analysis. The preliminary biological activity showed that some compounds possessed insecticidal activities against Culex pipiens pallens and Leucania separata Walk. The mortality ratio of some compounds against C. pipiens pallens coqilletti were 95% ~ 100% at concentration of 50 mg/L. The mortality ratio of e against L. separata was 100% at 1 000 mg/L.

**Key words** acylation; hydrazone; synthesis; bioactivity

腙基团 (—C=N—N—) 是医药和农药化合物中的重要结构, 不少含腙结构的化合物具有抑菌、消炎止痛、抗肿瘤和治疗肺结核等作用<sup>[1~4]</sup>。1973 年杜邦公司首次介绍了某些结构简单的二苯

甲酮腙的杀虫活性, 其后此类化合物引起了众多农药公司与科研工作者的兴趣。含腙结构的化合物作为农药具有制备简单、活性优良、作用谱广、毒性小等优点, 目前已商品化的品种有伏蚁腙

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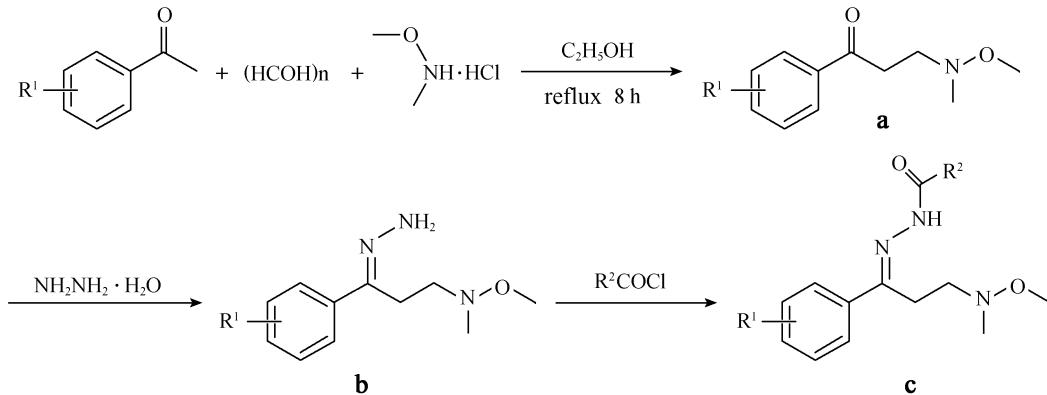
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(hydran ethanon)、嘧菌腙(ferm zone)和氟吡草腙(diflufenzopyr)等<sup>[5]</sup>。作者等曾经制备了取代苯丙酮肟衍生物,部分具有很好的杀菌活性<sup>[6]</sup>。在此基础上保留其基本化学结构,将其中的肟酰演变为腙进行结构改造,并通过酰化反应引入烷基、

取代芳基等,合成了一系列未见文献报道的N-酰基-3-(N-甲基-N-甲氧基氨基)-1-芳基-1-丙酮腙化合物(c),其结构经元素分析和核磁共振氢谱确证,并初步测试了其生物活性。

合成路线如下:



## 1 合成实验

### 1.1 仪器与试剂

Bruker AC 200型核磁共振仪(TM S为内标,CDCl<sub>3</sub>为溶剂); Carlo Erba EA 1110元素分析仪; WRS-1A型数字熔点仪(温度计未经校正); 100~200目硅胶; 试剂均为化学纯或分析纯。

### 1.2 化合物的制备

1.2.1 化合物a和b的制备 分别参照文献[7,8]方法合成。

1.2.2 目标化合物c的制备 以N-环丙酰基-3-(N-甲氧基-N-甲基氨基)-1-(4-丁基苯基)-1-丙酮腙(c<sub>1</sub>)的制备为例。

在100 mL三口烧瓶中加入0.79 g(0.003 mol)3-(N-甲氧基-N-甲基氨基)-1-(4-丁基苯基)-1-丙酮腙,0.30 g(0.003 mol)三乙胺,20 mL二氯甲烷。冰水浴冷却至0°C,搅拌下滴加0.31 g(0.003 mol)环丙酰氯。室温下搅拌反应4 h。用NaCl饱和溶液洗涤,无水MgSO<sub>4</sub>干燥,减压脱溶,经柱层析(石油醚-乙酸乙酯=2:1,体积比)纯化得白色粉末(c<sub>1</sub>),mp 129~130°C,收率74%。

采用同样的方法制得c<sub>2</sub>~c<sub>14</sub>,均为白色固体。所有目标化合物的熔点、元素分析和<sup>1</sup>H NMR数据分别见表1和表2。

Table 1 Physical constants and elemental analysis data of synthesized compounds c

| Com pd         | R <sup>1</sup>   | R <sup>2</sup>                                  | M p /C  | Y iel (%) | E lemental analysis(C alcd, %) |            |              |
|----------------|--|---|---------|-----------|--------------------------------|------------|--------------|
|                |  |   |         |           | C                              | H          | N            |
| c <sub>1</sub> | 4-(t-C <sub>4</sub> H <sub>9</sub> )-C <sub>6</sub> H <sub>4</sub> | cyclo-C <sub>3</sub> H <sub>5</sub>             | 129~130 | 74        | 68.73(68.85)                   | 8.85(8.82) | 12.62(12.68) |
| c <sub>2</sub> | 4-(t-C <sub>4</sub> H <sub>9</sub> )-C <sub>6</sub> H <sub>4</sub> | p-C <sub>6</sub> H <sub>4</sub>                 | 136~142 | 82        | 65.72(65.74)                   | 7.09(7.02) | 10.43(10.45) |
| c <sub>3</sub> | 4-(t-C <sub>4</sub> H <sub>9</sub> )-C <sub>6</sub> H <sub>4</sub> | 2-furan   | 104~105 | 79        | 67.12(67.20)                   | 7.58(7.61) | 11.81(11.76) |
| c <sub>4</sub> | 4-(t-C <sub>4</sub> H <sub>9</sub> )-C <sub>6</sub> H <sub>4</sub> | CH <sub>2</sub> CH <sub>2</sub> Cl              | 99~105  | 72        | 61.13(61.09)                   | 8.04(7.97) | 11.78(11.87) |
| c <sub>5</sub> | 4-(t-C <sub>4</sub> H <sub>9</sub> )-C <sub>6</sub> H <sub>4</sub> | o-CF <sub>3</sub> C <sub>6</sub> H <sub>4</sub> | 101~103 | 65        | 63.42(63.43)                   | 6.44(6.48) | 9.81(9.65)   |
| c <sub>6</sub> | 4-(t-C <sub>4</sub> H <sub>9</sub> )-C <sub>6</sub> H <sub>4</sub> | CF <sub>3</sub>                                 | 80~81   | 63        | 56.79(56.81)                   | 6.70(6.73) | 11.75(11.69) |
| c <sub>7</sub> | 2-naphthene  | i-C <sub>3</sub> H <sub>7</sub>                 | 84~85   | 68        | 55.13(55.10)                   | 7.51(7.47) | 14.80(14.83) |
| c <sub>8</sub> | 4-(t-C <sub>4</sub> H <sub>9</sub> )-C <sub>6</sub> H <sub>4</sub> | i-C <sub>3</sub> H <sub>7</sub>                 | 140~141 | 71        | 68.49(68.43)                   | 9.40(9.37) | 12.66(12.6)  |
| c <sub>9</sub> | 4-(t-C <sub>4</sub> H <sub>9</sub> )-C <sub>6</sub> H <sub>4</sub> | CH <sub>2</sub> CH <sub>3</sub>                 | 84~85   | 63        | 67.62(67.68)                   | 9.14(9.15) | 13.09(13.15) |

Continued

| Com pd          | R <sup>1</sup>   | R <sup>2</sup>   | Mp /C   | Yield (%) | Elemental analysis(Calcd, %) |            |              |
|-----------------|--|--|---------|-----------|------------------------------|------------|--------------|
|                 |  |  |         |           | C                            | H          | N            |
| c <sub>10</sub> | 2-thiophene  | CH <sub>2</sub> CH <sub>3</sub>                                    | 70~71   | 70        | 53.50(53.51)                 | 7.15(7.11) | 15.49(15.60) |
| c <sub>11</sub> | 2-thiophene  | p-(t-C <sub>4</sub> H <sub>9</sub> )-C <sub>6</sub> H <sub>4</sub> | 109~111 | 58        | 64.38(64.31)                 | 7.27(7.29) | 11.26(11.25) |
| c <sub>12</sub> | 2-thiophene  | m,p-C <sub>1</sub> C <sub>6</sub> H <sub>3</sub>                   | 130~132 | 52        | 49.79(49.75)                 | 4.40(4.44) | 10.85(10.88) |
| c <sub>13</sub> | 4-(C <sub>2</sub> H <sub>5</sub> O)-C <sub>6</sub> H <sub>4</sub>  | i-C <sub>3</sub> H <sub>7</sub>                                    | 120~121 | 80        | 63.50(63.53)                 | 8.49(8.47) | 13.02(13.07) |
| c <sub>14</sub> | 3,4-(CH <sub>3</sub> ) <sub>2</sub> -C <sub>6</sub> H <sub>4</sub> | i-C <sub>3</sub> H <sub>7</sub>                                    | 91~92   | 73        | 66.84(66.85)                 | 8.88(8.91) | 13.82(13.76) |

Table 2 <sup>1</sup>H NMR data of synthesized compounds c

| Com pd          | <sup>1</sup> H NMR(CDCl <sub>3</sub> /TMS), δ  |  |  |  |  |  |  |
|-----------------|--|--|--|--|--|--|--|
| c <sub>1</sub>  | 0.89~1.09(m, 4H, 2CH <sub>2</sub> ), 1.33[s, 9H, C(CH <sub>3</sub> ) <sub>3</sub> ], 2.62(s, 3H, NCH <sub>3</sub> ), 2.68~2.73(m, 1H, CH), 2.85~2.94(m, 4H, 2CH <sub>2</sub> ), 3.73(s, 3H, OCH <sub>3</sub> ), 7.45(d, 2H, J=8.52 Hz, A <sup>t</sup> H-3, 5), 7.94(d, 2H, J=8.52 Hz, A <sup>t</sup> H-2, 6), 9.92(s, 1H, NH)  |  |  |  |  |  |  |
| c <sub>2</sub>  | 1.29[s, 9H, C(CH <sub>3</sub> ) <sub>3</sub> ], 2.08(s, 3H, NCH <sub>3</sub> ), 2.83~2.86(m, 2H, CH <sub>2</sub> ), 3.07~3.13(m, 2H, CH <sub>2</sub> ), 3.38(s, 3H, OCH <sub>3</sub> ), 7.59(d, 2H, J=7.88 Hz, A <sup>t</sup> H-3, 5), 7.63(d, 2H, J=8.42 Hz, A <sup>t</sup> H-3', 5'), 7.79(d, 2H, J=8.42 Hz, A <sup>t</sup> H-2, 6), 7.76(d, 2H, J=8.42 Hz, A <sup>t</sup> H-2', 6'), 10.15(s, 1H, NH)   |  |  |  |  |  |  |
| c <sub>3</sub>  | 1.33[s, 9H, C(CH <sub>3</sub> ) <sub>3</sub> ], 2.69(s, 3H, NCH <sub>3</sub> ), 3.01~3.06(m, 4H, 2CH <sub>2</sub> ), 3.63(s, 3H, OCH <sub>3</sub> ), 6.56~7.44(m, 3H, Furan-H), 7.51(d, 2H, J=8.52 Hz, A <sup>t</sup> H-3, 5), 7.78(d, 2H, J=8.52 Hz, A <sup>t</sup> H-2, 6), 9.89(s, 1H, NH)  |  |  |  |  |  |  |
| c <sub>4</sub>  | 1.34[s, 9H, C(CH <sub>3</sub> ) <sub>3</sub> ], 3.63(s, 3H, NCH <sub>3</sub> ), 2.85~2.97(m, 4H, 2CH <sub>2</sub> ), 3.28(t, J=6.94 Hz, 2H, CH <sub>2</sub> CO), 3.68(s, 3H, OCH <sub>3</sub> ), 3.91(t, J=6.94 Hz, 2H, CH <sub>2</sub> CO), 7.43(d, 2H, J=8.50 Hz, A <sup>t</sup> H-3, 5), 7.68(d, 2H, J=8.50 Hz, A <sup>t</sup> H-2, 6), 10.15(s, 1H, NH)  |  |  |  |  |  |  |
| c <sub>5</sub>  | 1.34[s, 9H, C(CH <sub>3</sub> ) <sub>3</sub> ], 2.69(s, 3H, NCH <sub>3</sub> ), 3.03~3.05(m, 4H, 2CH <sub>2</sub> ), 3.41(s, 3H, OCH <sub>3</sub> ), 7.43(d, 2H, J=7.90 Hz, A <sup>t</sup> H-3, 5), 7.68(d, 2H, J=7.90 Hz, A <sup>t</sup> H-2, 6), 7.78~8.16(m, 4H, A <sup>t</sup> H-2', 3', 4', 6'), 10.15(s, 1H, NH)   |  |  |  |  |  |  |
| c <sub>6</sub>  | 1.34[s, 9H, C(CH <sub>3</sub> ) <sub>3</sub> ], 2.71(s, 3H, NCH <sub>3</sub> ), 2.64~2.70(m, 2H, CH <sub>2</sub> ), 2.83~2.94(m, 2H, CH <sub>2</sub> ), 3.60(s, 3H, OCH <sub>3</sub> ), 7.44(d, 2H, J=7.92 Hz, A <sup>t</sup> H-3, 5), 7.74(d, 2H, J=7.92 Hz, A <sup>t</sup> H-2, 6), 12.42(s, 1H, NH)   |  |  |  |  |  |  |
| c <sub>7</sub>  | 1.22[d, 6H, J=6.38 Hz, C(CH <sub>3</sub> ) <sub>2</sub> ], 2.62(s, 3H, NCH <sub>3</sub> ), 2.86~2.97(m, 4H, 2CH <sub>2</sub> ), 3.47~3.49(m, 1H, CH), 3.68(s, 3H, OCH <sub>3</sub> ), 7.02(d, d, 1H, J <sub>1</sub> =3.42, J <sub>2</sub> =4.60 Hz, ThiophenH-4), 7.24(d, d, 1H, J <sub>1</sub> =3.42, J <sub>2</sub> =1.23 Hz, ThiophenH-5), 7.31(d, d, 1H, J <sub>1</sub> =4.60, J <sub>2</sub> =1.23 Hz, ThiophenH-3), 9.78(s, 1H, NH)  |  |  |  |  |  |  |
| c <sub>8</sub>  | 1.19[d, 6H, J=6.48 Hz, C(CH <sub>3</sub> ) <sub>2</sub> ], 1.31[s, 9H, C(CH <sub>3</sub> ) <sub>3</sub> ], 2.64(s, 3H, NCH <sub>3</sub> ), 2.83~2.86(m, 2H, CH <sub>2</sub> ), 2.91~2.94(m, 2H, CH <sub>2</sub> ), 3.63~3.65(m, 1H, CH), 3.69(s, 3H, OCH <sub>3</sub> ), 7.41(d, 2H, J=8.48 Hz, A <sup>t</sup> H-3, 5), 7.69(d, 2H, J=8.48 Hz, A <sup>t</sup> H-2, 6), 9.59(s, 1H, NH)   |  |  |  |  |  |  |
| c <sub>9</sub>  | 1.22(t, J=6.72 Hz, 3H, CH <sub>3</sub> ), 1.33[s, 9H, C(CH <sub>3</sub> ) <sub>3</sub> ], 2.61(s, 3H, NCH <sub>3</sub> ), 2.70~2.82(m, 4H, 2CH <sub>2</sub> ), 2.85~2.93(m, 2H, CH <sub>2</sub> CO), 3.64(s, 3H, OCH <sub>3</sub> ), 7.41(d, 2H, J=8.50 Hz, A <sup>t</sup> H-3, 5), 7.68(d, 2H, J=8.50 Hz, A <sup>t</sup> H-2, 6), 9.70(s, 1H, NH)   |  |  |  |  |  |  |
| c <sub>10</sub> | 1.21(t, J=6.75 Hz, 3H, CH <sub>3</sub> ), 2.63(s, 3H, NCH <sub>3</sub> ), 2.62~2.72(m, 2H, CH <sub>2</sub> ), 2.80~2.88(m, 2H, CH <sub>2</sub> ), 2.90~2.92(m, 2H, CH <sub>2</sub> CO), 3.64(s, 3H, OCH <sub>3</sub> ), 7.02(d, d, 1H, J <sub>1</sub> =3.36, J <sub>2</sub> =4.63 Hz, ThiophenH-4), 7.23(d, d, 1H, J <sub>1</sub> =3.36, J <sub>2</sub> =1.21 Hz, 1H, ThiophenH-5), 7.31(d, d, 1H, J <sub>1</sub> =4.63, J <sub>2</sub> =1.21 Hz, 1H, ThiophenH-3), 9.84(s, 1H, NH)                        |  |  |  |  |  |  |
| c <sub>11</sub> | 1.33[s, 9H, C(CH <sub>3</sub> ) <sub>3</sub> ], 2.67(s, 3H, NCH <sub>3</sub> ), 2.95~3.09(m, 4H, 2CH <sub>2</sub> ), 3.46(s, 3H, OCH <sub>3</sub> ), 7.03(d, d, 1H, J <sub>1</sub> =3.40, J <sub>2</sub> =4.61 Hz, ThiophenH-4), 7.31(d, 2H, J=8.48 Hz, A <sup>t</sup> H-3', 5'), 7.45(d, d, 1H, J <sub>1</sub> =3.40, J <sub>2</sub> =1.21 Hz, ThiophenH-5), 7.49(d, d, 1H, J <sub>1</sub> =4.61, J <sub>2</sub> =1.21 Hz, ThiophenH-3), 7.83(d, 2H, J=8.48 Hz, A <sup>t</sup> H-2', 6'), 9.78(s, 1H, NH) |  |  |  |  |  |  |
| c <sub>12</sub> | 2.66(s, 3H, NCH <sub>3</sub> ), 2.86~3.01(m, 4H, 2CH <sub>2</sub> ), 3.73(s, 3H, OCH <sub>3</sub> ), 6.97(d, d, 1H, J <sub>1</sub> =3.45, J <sub>2</sub> =4.56 Hz, ThiophenH-4), 7.24(d, d, 1H, J <sub>1</sub> =3.45, J <sub>2</sub> =1.24 Hz, ThiophenH-5), 7.31(d, d, 1H, J <sub>1</sub> =4.56, J <sub>2</sub> =1.24 Hz, ThiophenH-3), 7.33~7.58(m, 3H, A <sup>t</sup> H-3', 5', 6'), 9.72(s, 1H, NH)  |  |  |  |  |  |  |
| c <sub>13</sub> | 1.25[d, 6H, J=6.49 Hz, C(CH <sub>3</sub> ) <sub>2</sub> ], 1.44(t, J=7.42 Hz, 3H, CH <sub>3</sub> ), 2.62(s, 3H, NCH <sub>3</sub> ), 2.76~2.93(m, 4H, 2CH <sub>2</sub> ), 2.89~2.96(m, 2H, CH <sub>2</sub> ), 3.68(s, 3H, OCH <sub>3</sub> ), 4.09~4.15(m, 2H, CH <sub>2</sub> ), 6.89(d, 2H, J=8.50 Hz, A <sup>t</sup> H-3, 5), 7.74(d, 2H, J=8.50 Hz, A <sup>t</sup> H-2, 6), 9.53(s, 1H, NH)  |  |  |  |  |  |  |
| c <sub>14</sub> | 1.26[d, 6H, J=6.52 Hz, C(CH <sub>3</sub> ) <sub>2</sub> ], 2.31(s, 6H, 2CH <sub>3</sub> ), 2.65(s, 3H, NCH <sub>3</sub> ), 2.78~2.85(m, 2H, CH <sub>2</sub> ), 2.90~2.96(m, 2H, CH <sub>2</sub> ), 3.49~3.57(m, 1H, CH), 3.64(s, 3H, OCH <sub>3</sub> ), 7.17~7.49(m, 3H, A <sup>t</sup> H-3, 5, 6), 9.60(s, 1H, NH)   |  |  |  |  |  |  |

## 2 生物活性测定

### 2.1 杀虫活性

2.1.1 杀淡色库蚊 *Culex pipiens pallens* 活性 参照文献[9]方法测定。

2.1.2 杀粘虫活性 采用国家南方农药创制中心生测标准程序之 SOP-叶片浸渍法测定。从温室剪下 2~4 叶期的玉米植株, 在稀释好的药液中充分浸渍 5 s 取出后悬挂在通风柜中; 放置 2~3 h 后, 剪下 30 mm 左右长的叶段, 置于小试管中, 每支试管约放 6~8 片叶段。用毛笔取 10 只 3 龄粘虫 *Leucania separata* 幼虫放入试管中, 纱布封口, 以不含药剂的溶剂为空白对照。以触动虫体不能正常爬行为死亡标准, 非正常的个体(爬行不自然、

半死、全死等)均计为死亡个体。

### 2.2 结果与讨论

该类化合物具有一定的杀虫活性(见表 3), 其中化合物 c<sub>1</sub>、c<sub>3</sub>、c<sub>8</sub> 和 c<sub>9</sub> 在 50 mg/L 时对淡色库蚊的致死率达到 100%, c<sub>2</sub> 和 c<sub>6</sub> 达到 95% 以上; 化合物 c<sub>1</sub> 在 1 000 mg/L 时对粘虫的致死率达到 100%。化合物对淡色库蚊的活性与 R<sup>1</sup> 为对叔丁基苯基有关, 这符合许多杀虫剂的特点, 如哒螨灵、虫酰肼等; 同时 R<sup>2</sup> 为环丙基或异丙基时, 化合物表现出较好的生物活性。

此外, 对化合物还进行了杀菌和除草活性的印鉴测试, 但均无活性。

Table 3 Effects of compounds c on different kinds of insect

| Compound       | Mortality(%)   |  | Compound        | Mortality(%)   |  |
|----------------|--|--|-----------------|--|--|
|                | <i>Culex pipiens pallens</i> <sup>*</sup><br>(50 mg/L) | <i>Leucania separata</i> <sup>**</sup><br>(1 000 mg/L) |                 | <i>Culex pipiens pallens</i> <sup>*</sup><br>(50 mg/L) | <i>Leucania separata</i> <sup>**</sup><br>(1 000 mg/L) |
| c <sub>1</sub> | 100  | 0  | c <sub>8</sub>  | 100  | 0  |
| c <sub>2</sub> | 97.3   | 0  | c <sub>9</sub>  | 100  | 0  |
| c <sub>3</sub> | 100  | 0  | c <sub>10</sub> | 0  | 79.2   |
| c <sub>4</sub> | 52.5   | 0  | c <sub>11</sub> | 0  | 0  |
| c <sub>5</sub> | 87.9   | 0  | c <sub>12</sub> | 0  | 0  |
| c <sub>6</sub> | 96.8   | 0  | c <sub>13</sub> | 85.7   | 0  |
| c <sub>7</sub> | 0  | 100  | c <sub>14</sub> | 0  | 50.0   |

\* n = 4 \*\* n = 3.

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