

# *N*-(1,5-二芳基-3-戊酮-1-基)-4-氨基苯甲酸的合成与 $\alpha$ -葡萄糖苷酶抑制活性初步研究

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**摘要:** 由 Mannich 反应直接合成了 16 个  $\beta$ -氨基酮化合物, 制备方法简单、反应条件温和、产品易纯化, 收率达到 45%~90%。所制备的化合物通过 IR、ESI-MS、<sup>1</sup>H NMR、<sup>13</sup>C NMR 和 HR-MS 等方法进行了结构表征。 $\alpha$ -葡萄糖苷酶抑制活性检测表明, 所合成的化合物对该酶的抑制活性顺序为: **2c** 最强, **2b**、**2h**、**1a**、**1f** 有一定的抑制活性。在此基础上, 对合成化合物的构-效关系进行了讨论。

**关键词:** 糖尿病;  $\alpha$ -葡萄糖苷酶;  $\beta$ -氨基酮化合物; Mannich 反应

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## Synthesis and $\alpha$ -glucosidase inhibitory activity of *N*-(1,5-diaryl-3-pentone-1-yl)-4-aminobenzoic acid

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**Abstract:** In order to find highly active antidiabetic lead compound, sixteen 4-aminobenzoic acid derivatives were designed and synthesized directly through Mannich reaction in the solution of ethanol at 15–35 °C with facile method, mild reaction condition and high yield (45%–90%). Fifteen of them are new compounds. Their structures were confirmed by <sup>1</sup>H NMR, <sup>13</sup>C NMR, IR, ESI-MS and HR-MS.  $\alpha$ -Glucosidase inhibitory activity of these compounds indicated that most of these compounds possess the activity with the order: **2c** > **2b** > **2h** > **1a** > **1f**. The structure-activity relationship of these 4-aminobenzoic acid derivatives was also discussed.

**Key words:** diabetes mellitus;  $\alpha$ -glucosidase;  $\beta$ -amino ketone; Mannich reaction

糖尿病作为全球第三大非传染性疾病已经受到广泛的关注。目前抗 II 型 (非胰岛素依赖型) 糖尿病药物分为四大类<sup>[1]</sup>: 磺酰脲类和非磺酰脲类促胰岛素分泌药物<sup>[2~5]</sup>, 双胍类、噻唑烷二酮类等胰岛素增敏剂<sup>[6~10]</sup>, 肝糖生成抑制剂<sup>[11]</sup>和  $\alpha$ -葡萄糖苷酶抑制剂等减缓糖吸收药物<sup>[12~14]</sup>。 $\alpha$ -葡萄糖苷酶抑制剂类药物对 I、II 型糖尿病均适用, 其作用机制为竞争性抑制小肠黏膜上皮细胞表面的  $\alpha$ -葡萄糖苷酶, 从而延缓碳水化合物的分解和吸收, 缓解餐后高血糖症, 具有广泛的应用前

景。然而, 由于该类药物使用时伴随腹泻、胃肠胀气等不良反应以及与其他降糖药联合使用时易引起低血糖<sup>[15]</sup>, 这使得寻找新型、副作用小的  $\alpha$ -葡萄糖苷酶抑制剂成为迫切需要。

为了发现新型  $\alpha$ -葡萄糖苷酶抑制剂, 必须借鉴已有药物的结构特点, 设计新型的抗糖尿病药物分子, 选择易于实现的合成方法。目前, 抗糖尿病药物种类很多, 结构类型各异。苯基丙酸类的 nateglinide (A4166)<sup>[4]</sup>、苯甲酸类型的 meglitinide 和 repaglinide<sup>[16]</sup>、 $\alpha$ -烷氧- $\beta$ -苯基丙酸的衍生物<sup>[17]</sup>、*L*-酪氨酸类衍生物<sup>[18]</sup>等分子中都含有羧基、氨基, 显示了这两种官能团的重要性; 另外这些分子多数都含有 2 个芳香环, 芳环之间间隔一定距离, 也就是说, 需要一个分子/结构

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单元 (linker) 将两个芳香环连接起来。Mannich 反应既是构筑有机分子骨架的重要方法, 也是将氮元素引入化合物的简便途径<sup>[19-22]</sup>, 更重要的是, 该反应可以实现三组分之间的一步联结。某些 Mannich 反应的产物 ( $\beta$ -氨基酮), 具有抗炎<sup>[23, 24]</sup>、抗癌<sup>[24-28]</sup>、抗菌<sup>[29-31]</sup>等生物活性。作者研究发现, 某些  $\beta$ -氨基酮衍生物还具有抗雄激素活性<sup>[32]</sup>, 这提示我们, 该类分子很有可能具有其他重要生物活性, 值得进一步研究。基于上述设想, 作者试图将对氨基苯甲酸 (类似于苯甲酸类抗糖尿病药物的药效团)、4-苯基-2-丁酮 (具有止咳的药理作用<sup>[33]</sup>) 或萘丁美酮 (具有消炎作用的上市药品) 作为原料, 通过 Mannich 反应合成新型的  $\beta$ -氨基酮衍生物, 以期获得具有新生物活性尤其是抗糖尿病活性的新化合物。合成实验发现, 4-苯基-2-丁酮或对氨基苯甲酸和芳香醛在较为温和的实验条件下发生 Mannich 反应, 可以直接得到 16 个 Mannich 碱 (1a~2h), 其中 15 个为新化合物, 反应式如图 1。

反应收率为 40%~90%。它们的化学结构得到 IR、ESI-MS (表 1)、<sup>1</sup>H NMR (表 2)、<sup>13</sup>C NMR (表 3) 和 HR-MS (表 4) 的确证。将所得到的化合物, 在抗糖尿病体外筛选模型上进行生物活性筛选, 发现它们对  $\alpha$ -葡萄糖苷酶有抑制作用, 其中 2c 对  $\alpha$ -葡萄糖苷酶有明显的抑制作用, 1a、1f、2b、2h 也显示出了较好的抑制活性。由此发现  $\beta$ -氨基酮具有抗糖尿病活性, 为新型抗糖尿药物的研制发现了新的先导化合物。上述 Mannich 碱的其他生物活性尚在进一步研究中。

## 结果及讨论

合成 Mannich 碱的途径主要有两种: Mannich 反应合成法和间接合成法。Mannich 反应通常采用三组分一锅法<sup>[34, 35]</sup>, 也有人选用 Schiff 碱与酮的缩合法<sup>[36]</sup>。Mannich 碱间接合成法有迈克尔加成法<sup>[37]</sup>、胺盐酸盐法<sup>[38]</sup>、胺交换法<sup>[28]</sup>、酮交换法<sup>[39, 40]</sup>等。本实验采用将三组分一锅法和 Schiff 碱与酮的缩合相结合的方法, 合成简便, 收率较高。

## 1 实验结果

Mannich 碱 (1a~2h) 的实验结果及分析数据见表 1~表 4。

## 2 结果讨论

芳香醛参与的 Mannich 反应, 多采用三组分一锅法或 Schiff 碱与酮的缩合法。在本课题组前期研究的基础上<sup>[41-44]</sup>, 本文采用的方法是两段投料一锅反应法, 即在醛胺反应生成 Schiff 碱后, 直接加入酮组分与之反应。该法既省去了 Schiff 碱的分离步骤, 又避免了三组分一锅法难于判断反应完成程度的不便。

**2.1 反应影响因素** 甲醛参与的 Mannich 反应一般需加热回流, 作者前期研究的芳香醛参与的 Mannich 反应一般在 0~40 °C 条件下进行。在本实验考察范围内, 合成此类  $\beta$ -氨基酮衍生物的难易差别很大。本文系统研究了反应条件, 试用了乙醇、乙醇-三氯甲烷、甲醇和三氯甲烷 4 种溶剂体系, 试验了浓盐酸、磷酸、冰醋酸、碘、氯化铁和氯化铝 6 种催化剂, 发现在温度为 25~30 °C、催化剂为浓 HCl (0.1~0.2 mL)、在无水乙醇和乙醇-三氯甲烷体系中反应结果最好。温度太低 (0 °C 左右), 反应速率降低, 甚至不发生反应; 温度超过 45 °C, 生成的产物会缓慢分解。盐酸过多或过少既会影响反应速度也会降低收率, 有时还决定反应是否进行。反应时间不宜过长, 延长反应时间容易导致产物分解, 所以适时监测反应对提高反应收率意义重大。这和作者以前的研究结果一致<sup>[41-44]</sup>。醛组分芳环上电子云密度对反应影响较大。间氯苯甲醛和 3, 4-二氯苯甲醛与 4-苯基-2-丁酮、对氨基苯甲酸难于发生 Mannich 反应, 和萘丁美酮参与的 Mannich 反应虽然可以发生, 但反应收率较低; 但对氯苯甲醛参与的上述 Mannich 反应, 不仅顺利发生而且收率较高。苯甲醛、间甲苯甲醛和胡椒醛, 与对氯苯甲醛相似, 反应较快且收率较高。含强吸电子基团的芳香醛, 如间硝基苯甲醛和对硝基苯甲醛, 其反应难易取决于取代基所处位置: 间硝基者反应较易发生, 收率也较高; 与之相反, 对硝基苯甲醛与 4-苯基-

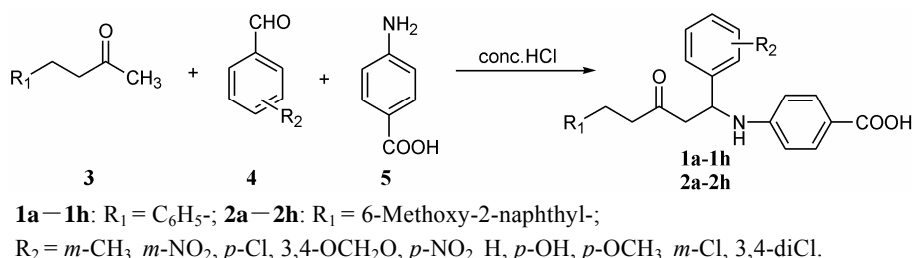


Figure 1 Mannich reaction of 4-aminobenzoic acid and aromatic aldehyde with 4-aryl-2-butanone

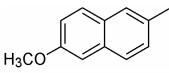
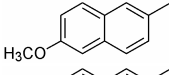
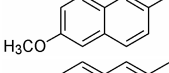
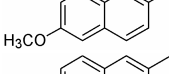
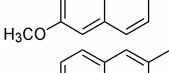
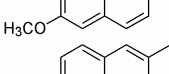
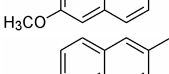
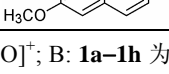
2-丁酮参与的反应收率较低,在萘丁美酮参与的反应中未能得到对应的 Mannich 碱。对甲氧基苯甲醛,不能发生反应或收率较低。幸运的是,含强供电子基团的对羟基苯甲醛,在本研究中都得到了对应的 Mannich 碱,收率达到 60%左右。

**2.2 生物活性结果分析** 生物活性结果见表 4。结果显示,在浓度为  $10 \mu\text{g}\cdot\text{mL}^{-1}$  时,合成的 16 个化合物大部分对  $\alpha$ -葡萄糖苷酶有抑制活性,其中 **2c** 有显著的抑制活性,抑制率达到了 66.5%,**2b**、**2h**、**1a** 和 **1f** 也达到了 34.3%~48.5%;在浓度为  $100 \mu\text{g}\cdot\text{mL}^{-1}$  时,测试过的化合物几乎都有显著的抑制活性。

分子结构模拟发现,1-芳基甲基酮、芳香醛和芳香胺生成的 Mannich 碱,分子刚性很大,三个芳环不在同一个平面;4-芳基-2-丁酮(4-苯基-2-丁酮和萘丁美酮)作为活性氢组分与芳香醛和芳香胺生成的 Mannich 碱,由于芳环和羰基之间多了两个亚甲基使分子有一定的柔韧性,体现在对  $\alpha$ -葡萄糖苷酶抑制活性上,4-芳基-2-丁酮型 Mannich 碱(后者)较 1-芳基甲基酮型(前者)的活性更好(前者活性另文发表);

即便是后者,结构虽然相似,但也有明显的活性差异,如醛组分相同,活性强度却是 **2c** (66.5%) > **1c** (-0.8%), **2h** (41.1%) > **1h** (6.6%),表现为萘丁美酮系列的 Mannich 碱的生物活性总体较 4-苯基-2-丁酮系列的更好,这可能与其空间结构的匹配性有关(**2c**、**2h** 与阿卡波糖的三维空间结构见图 2)。同一系列分子,当醛组分的取代基相同但取代位置不同时,抑制率也有明显差异,一般是对位取代产物的抑制活性优于间位,如: **2c** (*p*-Cl) > **2d** (*m*-Cl), **1a** (*p*-NO<sub>2</sub>) > **1b** (*m*-NO<sub>2</sub>)。对于取代位置相同但取代基不同的醛组分,所得产物的抑制率也不同,如: **2c**(*p*-Cl) > **2h**(*p*-OH) > **2e** (H), **2d** (*m*-Cl) > **2f** (*m*-CH<sub>3</sub>) > **2a** (*m*-NO<sub>2</sub>), **2b** (3, 4-diCl) > **2g** (3, 4-OCH<sub>2</sub>O), **1a** (*p*-NO<sub>2</sub>) > **1f** (*p*-OCH<sub>3</sub>) > **1d** (H) > **1h** (*p*-OH) > **1c** (*p*-Cl), **1b** (*m*-NO<sub>2</sub>) > **1e** (*m*-CH<sub>3</sub>)。由此可以看出,化合物的生物活性,既和取代基有关也和取代的位置相关。作者认为,生物活性的影响因素是多方面的(诸如电性、氢键、空间体积、分子刚柔性等),本文所涉及化合物的生物活性的规律性有待进一步研究。

**Table 1** Partial experimental results of target compounds

Compd.	R <sub>1</sub>	R <sub>2</sub>	Yield / %	mp / °C	MS ( <i>m/z</i> , %)			
					[M+1] <sup>+</sup>	[M+23] <sup>+</sup>	A	B
<b>1a</b>	-C <sub>6</sub> H <sub>5</sub>	<i>p</i> -NO <sub>2</sub>	53.3	148–149	419 (100)	441 (9)	133 (7)	120 (21)
<b>1b</b>	-C <sub>6</sub> H <sub>5</sub>	<i>m</i> -NO <sub>2</sub>	73.1	149–150	419 (100)	441 (17)	/	120 (25)
<b>1c</b>	-C <sub>6</sub> H <sub>5</sub>	<i>p</i> -Cl	67.2	162–163	408 (50)	430 (24)	133 (100)	120 (92)
<b>1d</b>	-C <sub>6</sub> H <sub>5</sub>	H	90.1	144–145	374 (65)	396 (25)	133 (67)	120 (100)
<b>1e</b>	-C <sub>6</sub> H <sub>5</sub>	<i>m</i> -CH <sub>3</sub>	83.0	163–164	388 (87)	410 (32)	133 (67)	120 (100)
<b>1f</b>	-C <sub>6</sub> H <sub>5</sub>	<i>p</i> -OCH <sub>3</sub>	64.2	154–155	402 (6) <sup>*</sup>	426 (43)	133 (100)	120 (51)
<b>1g</b>	-C <sub>6</sub> H <sub>5</sub>	3,4-OCH <sub>2</sub> O	80.4	160–161	418 (10)	440 (34)	133 (52)	120 (100)
<b>1h</b>	-C <sub>6</sub> H <sub>5</sub>	<i>p</i> -OH	60.5	125–127	388 (9) <sup>*</sup>	412 (100)	133 (37)	/
<b>2a</b>		<i>m</i> -NO <sub>2</sub>	60.1	156–159	499 (100)	521 (7)	213 (9)	271 (20)
<b>2b</b>		3,4-diCl	62.1	187–188	522 (100)	/	213 (44)	385 (34)
<b>2c</b>		<i>p</i> -Cl	84.4	191–194	488 (98)	510 (7)	213 (100)	351 (78)
<b>2d</b>		<i>m</i> -Cl	45.0	157–159	488 (100)	/	213 (33)	351 (12)
<b>2e</b>		H	72.3	193–194	454 (100)	476 (12)	213 (65)	317 (45)
<b>2f</b>		<i>m</i> -CH <sub>3</sub>	67.3	161–163	468 (100)	490 (10)	213 (47)	331 (64)
<b>2g</b>		3,4-OCH <sub>2</sub> O	89.0	191–192	496 (8) <sup>*</sup>	520 (25)	213 (47)	361 (27)
<b>2h</b>		<i>p</i> -OH	57.8	192–193	468 (5) <sup>*</sup>	492 (17)	213 (100)	333 (17)

A: [R<sub>1</sub>CH<sub>2</sub>CH<sub>2</sub>CO]<sup>+</sup>; B: **1a–1h** 为 [C<sub>6</sub>H<sub>4</sub>CO<sub>2</sub>H-p]<sup>+</sup>, **2a–2h** 为 [R<sub>1</sub>CH<sub>2</sub>CH<sub>2</sub>COCH<sub>2</sub>CHC<sub>6</sub>H<sub>4</sub>R<sub>2</sub>]<sup>+</sup>; \*为 [M-1]<sup>-</sup>

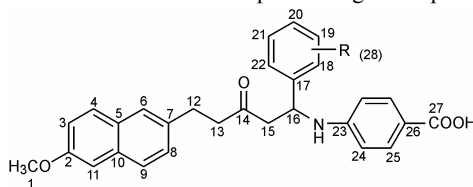
Table 2  $^1\text{H}$  NMR and IR data of target compounds

Compd.	$^1\text{H}$ NMR ( $\delta / 10^6$ , designated hydrogen atom)	IR ( $\nu_{\max}$ , $\text{cm}^{-1}$ )
<b>1a</b>	8.17 (2H, d, $J=8.6$ Hz, ArH), 7.67 (2H, d, $J=8.6$ Hz, ArH), 7.61 (2H, d, $J=8.6$ Hz, ArH), 7.25 (2H, d, $J=7.8$ Hz, ArH), 7.18~7.14 (3H, m, ArH), 6.56 (2H, d, $J=8.6$ Hz, ArH), 5.10 (1H, dd, $J=3.9, 8.4$ Hz, *CH), 3.08 (1H, dd, $J=7.8, 16.8$ Hz, $\text{CH}_2$ *CH), 2.88 (1H, dd, $J=4.3, 16.9$ Hz, $\text{CH}_2$ *CH), 2.83~2.77 (4H, m, $\text{CH}_2\text{CH}_2$ )	3 388(s), 3 000~ 2 500(br), 1 706(s), 1 661(s), 1 605(s)
<b>1b</b>	8.29 (1H, s, ArH), 8.08 (1H, d, $J=6.8$ Hz, ArH), 7.89~7.87 (1H, m, ArH), 7.62 (1H, d, $J=6.5$ Hz, ArH), 7.60 (2H, d, $J=8.6$ Hz, ArH), 7.22 (2H, d, $J=6.3$ Hz, ArH), 7.18~7.15 (3H, m, ArH), 6.60 (2H, d, $J=8.7$ Hz, ArH), 5.14 (1H, dd, $J=3.5, 8.3$ Hz, *CH), 3.09 (1H, dd, $J=7.9, 16.7$ Hz, $\text{CH}_2$ *CH), 2.89 (1H, dd, $J=4.9, 17.0$ Hz, $\text{CH}_2$ *CH), 2.84~2.75 (4H, m, $\text{CH}_2\text{CH}_2$ )	3 377(s), 3 000~ 2 500(br), 1 705(s), 1 660(s), 1 607(s), 1 501(s)
<b>1c</b>	7.60 (2H, d, $J=8.5$ Hz, ArH), 7.39 (2H, d, $J=8.0$ Hz, ArH), 7.37 (2H, d, $J=8.0$ Hz, ArH), 7.22 (2H, d, $J=6.9$ Hz, ArH), 7.18~7.15 (3H, m, ArH), 6.54 (2H, d, $J=8.6$ Hz, ArH), 4.95 (1H, dd, $J=3.8, 8.6$ Hz, *CH), 3.01 (1H, dd, $J=7.6, 16.5$ Hz, $\text{CH}_2$ *CH), 2.84~2.75 (5H, m, $\text{CH}_2\text{CH}_2$ , $\text{CH}_2$ *CH)	3 372(s), 3 000~ 2 500(br), 1 708(s), 1 669(s), 1 607(s), 1 496(s)
<b>1d</b>	7.59 (2H, d, $J=8.4$ Hz, ArH), 7.38 (2H, d, $J=7.4$ Hz, ArH), 7.30 (2H, d, $J=7.3$ Hz, ArH), 7.28~7.14 (6H, m, ArH), 6.55 (2H, d, $J=8.4$ Hz, ArH), 4.93 (1H, dd, $J=4.0, 8.6$ Hz, *CH), 3.02 (1H, dd, $J=7.3, 16.5$ Hz, $\text{CH}_2$ *CH), 2.84~2.75 (5H, m, $\text{CH}_2\text{CH}_2$ , $\text{CH}_2$ *CH)	3 394(s), 3 000~ 2 500(br), 1 710(s), 1 688(s), 1 605(s), 1 496(s)
<b>1e</b>	7.60 (2H, d, $J=8.7$ Hz, ArH), 7.26~7.14 (8H, m, ArH), 7.02 (1H, d, $J=6.8$ Hz, ArH), 6.54 (2H, d, $J=8.7$ Hz, ArH), 4.88 (1H, dd, $J=4.5, 8.7$ Hz, *CH), 3.00 (1H, dd, $J=7.2, 16.5$ Hz, $\text{CH}_2$ *CH), 2.86~2.74 (5H, m, $\text{CH}_2\text{CH}_2$ , $\text{CH}_2$ *CH), 2.26 (3H, s, $\text{CH}_3$ )	3 366(s), 3 000~ 2 500(br), 1 703(s), 1 673(s), 1 606(s), 1 498(s)
<b>1f</b>	7.59 (2H, d, $J=8.7$ Hz, ArH), 7.29 (2H, d, $J=8.7$ Hz, ArH), 7.22 (2H, d, $J=6.9$ Hz, ArH), 7.18~7.15 (3H, m, ArH), 6.85 (2H, d, $J=8.6$ Hz, ArH), 6.54 (2H, d, $J=8.7$ Hz, ArH), 4.87 (1H, dd, $J=3.2, 8.2$ Hz, *CH), 3.70 (3H, s, $\text{OCH}_3$ ), 2.99 (1H, dd, $J=7.3, 16.3$ Hz, $\text{CH}_2$ *CH), 2.83~2.72 (5H, m, $\text{CH}_2\text{CH}_2$ , $\text{CH}_2$ *CH)	3 393(s), 3 000~ 2 500(br), 1 710(s), 1 688(s), 1 604(s), 1 512(s)
<b>1g</b>	7.60 (2H, d, $J=8.5$ Hz, ArH), 7.24 (2H, d, $J=6.4$ Hz, ArH), 7.18~7.15 (3H, m, ArH), 6.95 (1H, s, ArH), 6.85~6.80 (2H, m, ArH), 6.55 (2H, d, $J=8.5$ Hz, ArH), 5.95 (2H, s, $\text{OCH}_2\text{O}$ ), 4.86 (1H, dd, $J=3.6, 8.1$ Hz, *CH), 2.97 (1H, dd, $J=7.2, 16.2$ Hz, $\text{CH}_2$ *CH), 2.87~2.74 (5H, m, $\text{CH}_2\text{CH}_2$ , $\text{CH}_2$ *CH)	3 394(s), 3 000~ 2 500(br), 1 710(s), 1 689(s), 1 603(s), 1 488(s)
<b>1h</b>	7.60 (2H, d, $J=8.5$ Hz, ArH), 7.22 (2H, d, $J=7.0$ Hz, ArH), 7.20~7.13 (5H, m, ArH), 6.68 (2H, d, $J=8.2$ Hz, ArH), 6.54 (2H, d, $J=8.5$ Hz, ArH), 4.82 (1H, dd, $J=3.2, 8.2$ Hz, *CH), 2.97 (1H, dd, $J=7.3, 16.2$ Hz, $\text{CH}_2$ *CH), 2.86~2.72 (5H, m, $\text{CH}_2\text{CH}_2$ , $\text{CH}_2$ *CH)	3 393(s), 3 000~ 2 500(br), 1 711(s), 1 690(s), 1 604(s), 1 513(s)
<b>2a</b>	8.30 (1H, s, ArH), 8.05 (1H, d, $J=7.4$ Hz, ArH), 7.89 (1H, d, $J=7.6$ Hz, ArH), 7.70~7.68 (2H, m, ArH), 7.67~7.65 (1H, m, ArH), 7.62 (2H, d, $J=8.7$ Hz, ArH), 7.57 (1H, s, ArH), 7.30 (1H, d, $J=7.9$ Hz, ArH), 7.26 (1H, s, ArH), 7.12 (1H, d, $J=7.2$ Hz, ArH), 6.60 (2H, d, $J=8.6$ Hz, ArH), 5.15 (1H, dd, $J=4.7, 8.4$ Hz, *CH), 3.85 (3H, s, $\text{OCH}_3$ ), 3.11 (1H, dd, $J=8.9, 16.6$ Hz, $\text{CH}_2$ *CH), 2.95~2.80 (5H, m, $\text{CH}_2\text{CH}_2$ , $\text{CH}_2$ *CH)	3 366(s), 3 000~ 2 500(br), 1 705(s), 1 676(s), 1 606(s), 1 504(s)
<b>2b</b>	7.72 (1H, s, ArH), 7.69~7.67 (3H, m, 2ArH, NH), 7.64 (2H, d, $J=8.7$ Hz, ArH), 7.57~7.54 (2H, m, ArH), 7.39 (1H, d, $J=7.9$ Hz, ArH), 7.31 (1H, d, $J=8.0$ Hz, ArH), 7.26 (1H, s, ArH), 7.12 (1H, d, $J=7.8$ Hz, ArH), 6.58 (2H, d, $J=8.6$ Hz, ArH), 5.00 (1H, dd, $J=4.8, 8.4$ Hz, *CH), 3.86 (3H, s, $\text{OCH}_3$ ), 3.06 (1H, dd, $J=9.2, 16.4$ Hz, $\text{CH}_2$ *CH), 2.97~2.85 (4H, m, $\text{CH}_2\text{CH}_2$ ), 2.83 (1H, dd, $J=4.6, 16.3$ Hz, $\text{CH}_2$ *CH)	3 384(s), 3 000~ 2 500(br), 1 708(s), 1 655(s), 1 607(s), 1 502(s)
<b>2c</b>	7.70 (1H, d, $J=7.7$ Hz, ArH), 7.68 (1H, d, $J=8.2$ Hz, ArH), 7.61 (2H, d, $J=8.6$ Hz, ArH), 7.55 (1H, s, ArH), 7.41 (2H, d, $J=8.4$ Hz, ArH), 7.34 (2H, d, $J=8.4$ Hz, ArH), 7.31 (1H, d, $J=9.0$ Hz, ArH), 7.27 (1H, s, ArH), 7.12 (1H, d, $J=8.9$ Hz, ArH), 6.55 (2H, d, $J=8.6$ Hz, ArH), 4.96 (1H, dd, $J=4.8, 8.5$ Hz, *CH), 3.85 (3H, s, $\text{OCH}_3$ ), 3.04 (1H, dd, $J=9.0, 16.5$ Hz, $\text{CH}_2$ *CH), 2.96~2.85 (4H, m, $\text{CH}_2\text{CH}_2$ ), 2.81 (1H, dd, $J=4.5, 16.4$ Hz, $\text{CH}_2$ *CH)	3 392(s), 3 000~ 2 500(br), 1 711(s), 1 691(s), 1 605(s), 1 490(s)
<b>2d</b>	7.71 (1H, d, $J=7.9$ Hz, ArH), 7.69 (1H, d, $J=7.4$ Hz, ArH), 7.61 (2H, d, $J=8.7$ Hz, ArH), 7.56 (1H, s, ArH), 7.46 (1H, s, ArH), 7.36 (1H, d, $J=7.6$ Hz, ArH), 7.33~7.27 (3H, m, ArH), 7.26 (1H, s, ArH), 7.11 (1H, d, $J=9.0$ Hz, ArH), 6.56 (2H, d, $J=8.6$ Hz, ArH), 4.98 (1H, dd, $J=4.2, 9.0$ Hz, *CH), 3.85 (3H, s, $\text{OCH}_3$ ), 3.04 (1H, dd, $J=8.8, 16.3$ Hz, $\text{CH}_2$ *CH), 2.95~2.84 (4H, m, $\text{CH}_2\text{CH}_2$ ), 2.83 (1H, dd, $J=4.9, 16.4$ Hz, $\text{CH}_2$ *CH)	3 385(s), 3 000~ 2 500(br), 1 708(s), 1 658(s), 1 605(s), 1 502(s)
<b>2e</b>	7.70~7.68 (2H, m, ArH), 7.61 (2H, d, $J=8.6$ Hz, ArH), 7.56 (1H, s, ArH), 7.39 (2H, d, $J=7.4$ Hz, ArH), 7.28~7.25 (4H, m, ArH), 7.20 (1H, d, $J=8.2$ Hz, ArH), 7.12 (1H, d, $J=8.3$ Hz, ArH), 6.57 (2H, d, $J=8.7$ Hz, ArH), 4.96 (1H, dd, $J=4.6, 8.9$ Hz, *CH), 3.85 (3H, s, $\text{OCH}_3$ ), 3.05 (1H, dd, $J=9.2, 16.4$ Hz, $\text{CH}_2$ *CH), 2.94~2.83 (4H, m, $\text{CH}_2\text{CH}_2$ ), 2.80 (1H, dd, $J=4.4, 16.2$ Hz, $\text{CH}_2$ *CH)	3 383(s), 3 000~ 2 500(br), 1 708(s), 1 656(s), 1 607(s), 1 502(s)
<b>2f</b>	7.70~7.68 (2H, m, ArH), 7.61 (2H, d, $J=8.6$ Hz, ArH), 7.55 (1H, s, ArH), 7.29 (1H, d, $J=8.2$ Hz, ArH), 7.27 (1H, s, ArH), 7.18~7.15 (3H, m, ArH), 7.13 (1H, d, $J=7.5$ Hz, ArH), 7.02 (1H, s, ArH), 6.56 (2H, d, $J=8.7$ Hz, ArH), 4.90 (1H, dd, $J=4.4, 8.8$ Hz, *CH), 3.85 (3H, s, $\text{OCH}_3$ ), 3.03 (1H, dd, $J=9.3, 16.4$ Hz, $\text{CH}_2$ *CH), 2.94~2.80 (4H, m, $\text{CH}_2\text{CH}_2$ ), 2.77 (1H, dd, $J=4.4, 16.4$ Hz, $\text{CH}_2$ *CH), 2.25 (3H, s, $\text{CH}_3$ )	3 394(s), 3 000~ 2 500(br), 1 711(s), 1 689(s), 1 603(s), 1 486(s)

Continued

Compd.	<sup>1</sup> H NMR (δ / 10 <sup>-6</sup> , designated hydrogen atom)	IR (ν <sub>max</sub> , cm <sup>-1</sup> )
<b>2g</b>	7.70~7.68 (2H, m, ArH), 7.61 (2H, d, J=8.2 Hz, ArH), 7.56 (1H, s, ArH), 7.29 (1H, d, J=8.2 Hz, ArH), 7.26 (1H, s, ArH), 7.11 (1H, d, J=8.9 Hz, ArH), 6.96 (1H, s, ArH), 6.85 (1H, d, J=8.0 Hz, ArH), 6.82 (1H, d, J=7.9 Hz, ArH), 6.57 (2H, d, J=8.3 Hz, ArH), 5.95 (2H, s, OCH <sub>2</sub> O), 4.88 (1H, dd, J=4.4, 8.9 Hz, *CH), 3.85 (3H, s, OCH <sub>3</sub> ), 3.00 (1H, dd, J=9.2, 16.3 Hz, <u>CH</u> <sub>2</sub> *CH), 2.93~2.80 (4H, m, CH <sub>2</sub> CH <sub>2</sub> ), 2.77 (1H, dd, J=4.5, 16.3 Hz, <u>CH</u> <sub>2</sub> *CH)	3 396(s), 3 000~ 2 500(br), 1 710(s), 1 688(s), 1 604(s), 1 486(s)
<b>2h</b>	7.71 (1H, d, J=7.1 Hz, ArH), 7.69 (1H, d, J=8.2 Hz, ArH), 7.61 (2H, d, J=8.6 Hz, ArH), 7.54 (1H, s, ArH), 7.28 (1H, d, J=8.1 Hz, ArH), 7.26 (1H, s, ArH), 7.18 (2H, d, J=8.4 Hz, ArH), 7.12 (1H, d, J=8.7 Hz, ArH), 6.69 (2H, d, J=8.4 Hz, ArH), 6.55 (2H, d, J=8.7 Hz, ArH), 4.84 (1H, dd, J=5.0, 8.4 Hz, *CH), 3.85 (3H, s, OCH <sub>3</sub> ), 3.00 (1H, dd, J=9.0, 16.2 Hz, <u>CH</u> <sub>2</sub> *CH), 2.92~2.79 (4H, m, CH <sub>2</sub> CH <sub>2</sub> ), 2.76 (1H, dd, J=4.9, 16.4 Hz, <u>CH</u> <sub>2</sub> *CH)	3 398(s), 3 000~ 2 500(br), 1 709(s), 1 659(s), 1 607(s), 1 509(s)

**Table 3** <sup>13</sup>C NMR data of partial target compounds



Compd.	<sup>13</sup> C NMR (δ/10 <sup>-6</sup> , designated carbon atom)
<b>2a</b>	206.9 (14), 167.5 (27), 156.9 (2), 151.2 (23), 148.1 (19), 145.9 (17), 136.3 (7), 133.7 (22), 132.9 (10), 131.2 (25), 130.1 (21), 128.9 (5), 128.7 (4), 127.8 (8), 126.8 (6), 126.0 (9), 122.2 (18), 121.3 (20), 118.6 (26), 118.0 (3), 112.0 (24), 105.8 (11), 55.3 (1), 51.6 (16), 49.7 (15), 43.5 (13), 28.9 (12)
<b>2b</b>	206.9 (14), 167.5 (27), 156.9 (2), 151.3 (23), 144.7 (17), 136.3 (7), 132.9 (19), 131.9 (10), 131.3 (25), 130.8 (20), 129.6 (21), 128.9 (5), 128.8 (4), 128.7 (8), 127.8 (18), 127.2 (6), 126.9 (9), 126.0 (22), 118.7 (26), 118.0 (3), 112.1 (24), 105.9 (11), 55.3 (1), 51.3 (16), 49.7 (15), 43.6 (13), 28.9 (12)
<b>2c</b>	207.1 (14), 167.5 (27), 157.0 (2), 151.4 (23), 142.3 (17), 132.9 (7), 131.6 (10), 131.1 (20), 130.3 (25), 130.2 (5), 129.0 (4), 128.9 (8), 128.7 (19, 21), 128.5 (18, 22), 127.8 (6), 126.0 (9), 118.6 (26), 117.8 (3), 112.0 (24), 105.9 (11), 55.3 (1), 51.8 (16), 50.0 (15), 43.7 (13), 28.9 (12)
<b>2d</b>	207.1 (14), 167.5 (27), 156.9 (2), 151.5 (23), 144.7 (17), 134.2 (7), 133.3 (19), 132.4 (10), 131.6 (25), 130.1 (21), 129.1 (5), 129.0 (4), 128.5 (8), 127.6 (20), 127.5 (18), 127.1 (6), 126.4 (9), 125.1 (22), 119.0 (26), 118.9 (3), 113.6 (24), 106.2 (11), 55.9 (1), 53.0 (16), 49.9 (15), 43.5 (13), 28.9 (12)
<b>2e</b>	207.3 (14), 167.6 (27), 156.9 (2), 151.7 (23), 143.3 (17), 136.3 (7), 132.9 (10), 131.2 (25), 129.0 (5), 128.8 (4), 128.6 (19, 21), 127.8 (8), 127.1 (20), 126.9 (6), 126.6 (18, 22), 126.0 (9), 118.6 (26), 117.5 (3), 111.9 (24), 105.9 (11), 55.3 (1), 52.4 (16), 50.3 (15), 43.7 (13), 28.9 (12)
<b>2f</b>	207.4 (14), 167.6 (27), 156.9 (2), 151.7 (23), 143.3 (17), 137.7 (19), 136.3 (7), 132.9 (10), 131.2 (25), 129.0 (5), 128.7 (4), 128.5 (18), 127.8 (8, 21), 127.1 (6), 126.9 (9), 126.0 (20), 123.7 (22), 118.6 (26), 117.4 (3), 111.9 (24), 105.9 (11), 55.3 (1), 52.4 (16), 50.3 (15), 43.7 (13), 28.9 (12), 21.3 (28)
<b>2g</b>	207.4 (14), 167.5 (27), 156.9 (2), 151.6 (23), 147.5 (19), 146.2 (20), 137.3 (17), 136.3 (7), 132.9 (10), 131.1 (25), 128.9 (5), 128.7 (4), 127.8 (8), 126.9 (6), 126.0 (9), 119.9 (22), 118.6 (26), 117.5 (3), 112.0 (21), 108.2 (24), 106.9 (18), 105.9 (11), 101.0 (28), 55.3 (1), 52.1 (16), 50.4 (15), 43.7 (13), 28.9 (12)
<b>2h</b>	207.6 (14), 167.6 (27), 157.0 (20), 156.3 (2), 151.8 (23), 136.4 (17), 133.4 (7), 132.9 (10), 131.2 (25), 129.0 (5), 128.7 (4), 127.8 (8), 127.7 (18, 22), 126.9 (6), 126.0 (9), 118.6 (26), 117.3 (3), 115.3 (19, 21), 112.0 (24), 105.9 (11), 55.3 (1), 52.0 (16), 50.5 (15), 43.8 (13), 29.0 (12)

**Table 4** The HR-MS and biological activity data of target compounds

Compd.	Formula	HR-MS		Conc.1		Inhibition /%	Conc.2		Inhibition /%
		Calcd	Found	μg·mL <sup>-1</sup>	nmol·mL <sup>-1</sup>		μg·mL <sup>-1</sup>	nmol·mL <sup>-1</sup>	
<b>1a</b>	C <sub>24</sub> H <sub>22</sub> N <sub>2</sub> NaO <sub>5</sub>	441.142 1	441.141 4	10	23.9	36.3	100	239.2	68.4
<b>1b</b>	C <sub>24</sub> H <sub>22</sub> N <sub>2</sub> NaO <sub>5</sub>	441.142 1	441.140 2	10	23.9	19.8	100	239.2	80.8
<b>1c</b>	C <sub>24</sub> H <sub>22</sub> ClN <sub>2</sub> NaO <sub>3</sub>	430.118 0	430.118 2	10	24.6	-0.8	100	245.7	82.4
<b>1d</b>	C <sub>24</sub> H <sub>23</sub> NNaO <sub>3</sub>	396.175 0	396.157 4	10	26.8	19.8	100	268.1	63.1
<b>1e</b>	C <sub>25</sub> H <sub>25</sub> NNaO <sub>3</sub>	410.172 7	410.171 9	10	25.8	14.8	100	258.4	37.8
<b>1f</b>	C <sub>25</sub> H <sub>25</sub> NNaO <sub>4</sub>	426.167 6	426.167 8	10	24.8	34.3	/	/	/
<b>1g</b>	C <sub>25</sub> H <sub>23</sub> NNaO <sub>5</sub>	440.146 8	440.145 3	10	23.9	-5.3	100	239.8	62.3
<b>1h</b>	C <sub>24</sub> H <sub>23</sub> NNaO <sub>4</sub>	412.151 9	412.152 9	10	25.7	6.6	100	257.1	70.8
<b>2a</b>	C <sub>29</sub> H <sub>26</sub> N <sub>2</sub> NaO <sub>6</sub>	521.168 3	521.167 2	10	20.1	6.4	/	/	/
<b>2b</b>	C <sub>29</sub> H <sub>25</sub> Cl <sub>2</sub> NNaO <sub>4</sub>	544.105 3	544.102 8	10	19.2	48.5	/	/	/
<b>2c</b>	C <sub>29</sub> H <sub>26</sub> ClNNaO <sub>4</sub>	510.144 3	510.143 0	10	20.5	66.5	/	/	/
<b>2d</b>	C <sub>29</sub> H <sub>26</sub> ClNNaO <sub>4</sub>	510.144 3	510.142 5	10	20.5	21.8	/	/	/
<b>2e</b>	C <sub>29</sub> H <sub>27</sub> NNaO <sub>4</sub>	476.183 2	476.185 1	10	22.1	10.2	/	/	/
<b>2f</b>	C <sub>30</sub> H <sub>29</sub> NNaO <sub>4</sub>	490.198 9	490.199 4	10	21.4	6.6	/	/	/
<b>2g</b>	C <sub>30</sub> H <sub>27</sub> NNaO <sub>6</sub>	520.173 1	520.173 5	10	20.1	7.9	/	/	/
<b>2h</b>	C <sub>29</sub> H <sub>27</sub> NNaO <sub>5</sub>	492.178 1	492.176 7	10	21.4	41.1	/	/	/
	Acarbose	645.57	/	100	154.9	74.1	100	154.9	74.1

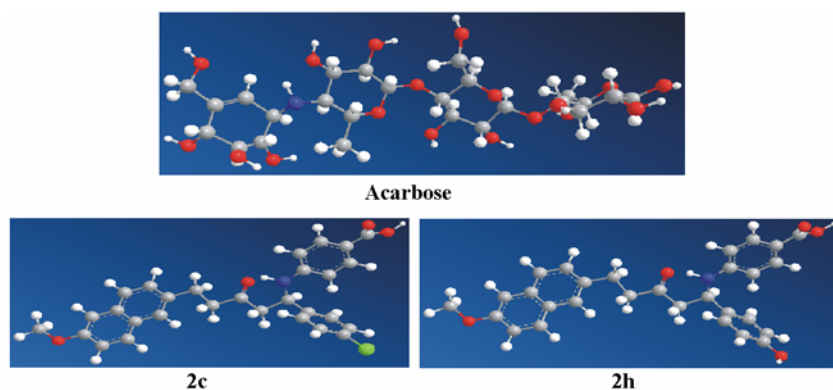


Figure 2 Three-dimensional structures of Acarbose and compounds 2c, 2h

## 结论

本文设计合成了 16 个 4-芳基-2-丁酮、芳香醛与对氨基苯甲酸生成的 Mannich 碱; 首次发现这些  $\beta$ -氨基酮化合物对  $\alpha$ -葡糖苷酶具有不同程度的抑制活性, 其中 2c 有显著的抑制活性, 可作为新型抗糖尿病药物先导分子进一步研究。

## 实验部分

### 1 主要实验仪器及试剂

精密显微熔点测定仪 (X-6 型, 北京福凯仪器有限公司); FT-IR 红外光谱仪 (GX 型, USA; KBr 压片); 核磁共振仪 (AV-300, USA; TMS 为内标, DMSO- $d_6$  为溶剂); 质谱仪 (Agilent 1946B ESI MS, USA); HR-MS (Bruker Daltonics Data Analysis 3.2, USA)。葡糖苷酶 (Sigma, G-0660), 葡萄糖检测试剂 (南京建成公司), 芳香醛、4-苯基-2-丁酮、蔡丁美酮和对氨基苯甲酸为国产化学纯, 其余为国产分析纯。

### 2 Mannich 碱 (1a~2h) 的一般合成方法

在 50 mL 圆底烧瓶中加入 2 mmol 的对氨基苯甲酸、2 mmol 芳香醛、2~4 mL 无水乙醇, 电磁搅拌。产生大量沉淀后, 加入 2 mmol 酮组分, 滴加 0.1~0.2 mL 浓盐酸, 继续搅拌。TLC 监测反应进程。反应完成后, 置于冰箱中过夜。次日抽滤, 滤饼水洗 (2×3 mL)、醇洗 (2×3 mL), 抽干得粗品。乙醇-丙酮混合溶剂重结晶, 干燥后得纯品。测定熔点, 进行 IR、 $^1\text{H}$  NMR、 $^{13}\text{C}$  NMR、ESI-MS 和 HR-MS 测定。

### 3 Mannich 碱的生物活性测定

100  $\mu\text{L}$  反应体系中含 0.02 U 葡糖苷酶、67  $\text{nmol}\cdot\text{mL}^{-1}$  磷酸钠缓冲液 (pH 6.8) 和样品, 同时设立空白对照 (不含酶和样品) 和阴性对照 (不含样品), 37  $^\circ\text{C}$  反应 10 min, 加入 0.1  $\text{mol}\cdot\text{mL}^{-1}$  麦芽糖, 室温反应 10 min, 再加入 200  $\mu\text{L}$  的葡萄糖检测试剂, 混匀后

490 nm 测定  $A$  值。根据  $A$  值计算抑制率, 抑制率 =  $[1 - (A_{\text{样品}} - A_{\text{空白}}) / (A_{\text{阴性}} - A_{\text{空白}})] \times 100$ 。每个样品每个浓度设双复孔, 重复两次。

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